

Desoxypatulinic Acid from a Patulin-Producing Strain of *Penicillium patulum*

Desoxypatulinic acid, previously obtained by synthesis¹ and by hydrogenation of patulin², has not been reported as a natural product. We have now isolated it from cultures of *Penicillium patulum* Bainier (*P. urticae* Bainier), a known source of many fungal metabolites, including patulin^{3,4}.

*P. patulum*⁵, strain No. 562 isolated from a mixed feed suspected of causing death of a number of cattle, was grown at room temperature in 2.8-liter Fernbach flasks each containing 200 ml of yeast extract (2%)–sucrose (15%) liquid medium. After 9–11 days incubation, the medium was extracted with 3 equal volumes of ethyl acetate. The solvent was evaporated, and the residue dissolved in warm benzene and chromatographed on a column of silica gel. Elution with benzene–ethyl acetate (9:1, v/v) yielded patulin (I) (124 mg per Fernbach flask), m.p. 110–111°C after recrystallization from benzene; it was identified by comparison with standard material (mixed m.p., UV- and IR-spectra, and thin-layer chromatography (TLC)). Further elution of the column with benzene–ethyl acetate (85:15, v/v) gave fractions containing griseofulvin (identified by comparative TLC). A more polar compound (80 mg crystalline material per Fernbach flask) was eluted by benzene–ethyl acetate (85:15 to 75:25, v/v). It was purified by crystallization from *iso*-propyl ether–*n*-hexane and toluene followed by sublimation. The crystals, m.p. 115–115.5°C, were soluble in water (to give an acidic pH) and aqueous NaHCO₃ (with effervescence). The molecular formula was C₇H₆O₄ (found: C, 53.94; H, 5.16. Calc. for C₇H₆O₄: C, 53.85; H, 5.16%). UV- and IR-spectral properties were λ_{max} (EtOH) 267 nm (ϵ 7960); ν_{max} (CHCl₃) 3510 (OH), bands between 3000 and 2400 (COOH), 1718 (COOH dimer), 1672 (conjugated C=O), 1621 (C=C) cm⁻¹. The 100 MHz NMR-spectrum (in CDCl₃) consisted of signals at τ 1.53 (singlet, 1P, disappears on addition of D₂O; OH), 2.63 (singlet, 1P; CO·C=CH·O), 5.48 (triplet, 2P, $J=7$ Hz; CH₂O), 6.83 (singlet, 2P; CO·CH₂·C=C), and 7.34 (triplet, 2P, $J=7$ Hz; CO·CH₂); each triplet collapsed to a singlet on irradiation at the frequency of the other triplet. The mass spectrum (70 e.v.) showed a parent ion at m/e 156 and other prominent ions at m/e 112 (M–CO₂), 83, 60 [CH₂=C(OH)₂]⁺, 55, 39 and 27⁶. Consideration of the

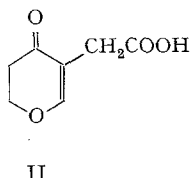
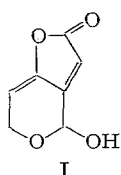
foregoing evidence led to formulation of the metabolite as 2, 3-dihydro-4-pyrone-5-acetic acid (II), desoxypatulinic acid [m.p. 114.5–115.5°C; λ_{max} (EtOH) 268 nm]^{1,2}. This assignment was confirmed by direct comparison of the metabolite (mixed m.p., TLC, and IR-, UV- and mass spectra) with desoxypatulinic acid, m.p. 113.5–114.5°C, prepared by hydrogenolysis of chlorodesoxypatulinic acid². Desoxypatulinic acid (46 µg/ml tryptic soy agar) had no inhibitory effect on *Bacillus megaterium*, *B. subtilis*, *B. cereus*, *Staphylococcus aureus*, *S. epidermidis*, *Sarcina lutea*, *Micrococcus flavus*, and *Saccharomyces cerevisiae* (cf. patulin was completely inhibitory under the same conditions).

Desoxypatulinic acid was readily detected by TLC on silica gel F-254 (0.25 mm) developed with toluene–ethyl acetate–formic acid (6:3:1, v/v) as a dark spot under short wavelength UV-light at R_f 0.22. Analysis by TLC of aliquots of culture medium withdrawn during the fermentation showed that desoxypatulinic acid and patulin (R_f 0.33) concentrations reached a peak at about the same time (10–12 days); patulin was no longer detectable on day 17 although traces of the acid remained. The relationship of desoxypatulinic acid to the pathway of patulin biosynthesis^{4,7,8} merits investigation.

Zusammenfassung. Aus Kulturlösungen von *Penicillium patulum* Bainier wurde neben Patulin die Desoxypatulin-säure isoliert und identifiziert.

P. M. SCOTT, B. KENNEDY and W. VAN WALBEEK

Food Research Laboratories,
Health Protection Branch,
Department of National Health and Welfare,
Ottawa (Ontario K1A 0L2, Canada), 28 February 1972.



Systematics of a *Leptospira* Strain Isolated from Frog

In the year 1964, in Iowa, USA, DIESCH et al.^{1,2} isolated a leptospira strain from a pool of kidneys of 3 specimen of frogs (*Rana pipiens*) collected during follow-up of a human leptospirosis outbreak associated with swimming.

This is the first case reported in the literature of isolation of a leptospira strain from the organs of an amphibian. Therefore, it is of particular interest, also for epidemiological purposes, to ascertain whether this strain belongs to pathogenic leptospires (*L. interrogans*) or to saprophytic ones ('biflexa' complex), and whether this strain might be

inserted in the group of leptospiral serotypes that we know already. DIESCH et al.³ have executed some surveys in this direction without being able to attain a sure conclusion. In fact, the leptospira, inoculated into guinea-

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- ² B. G. ENGEL, W. BRZESKI and P. A. PLATTNER, *Helv. chim. Acta* 32, 1166 (1949).
- ³ S. SHIBATA, S. NATORI and S. UDAGAWA, *List of Fungal Products* (Charles C. Thomas, Springfield, Illinois 1964).
- ⁴ E. W. BASSETT and S. W. TANENBAUM, *Experientia* 14, 38 (1958).
- ⁵ The fungus was identified at the Centraalbureau voor Schimmelcultures, Baarn, The Netherlands.
- ⁶ Analogous prominent ions at m/e 190 (M), 146 (M–CO₂), 83, 60, 55, 39 and 27 were observed, inter alia, in the mass spectrum of chlorodesoxypatulinic acid (3-chloro-2,3-dihydro-4-pyrone-5-acetic acid)¹, m.p. 132–132.5°C, prepared by the method of ENGEL et al. (1949)². We thank W. F. MILES for recording mass spectra.
- ⁷ S. W. TANENBAUM and E. W. BASSETT, *J. biol. Chem.* 234, 1861 (1959).
- ⁸ A. I. SCOTT and M. YALPANI, *Chem. Commun.* 1967, 945.

- ¹ S. L. DIESCH, W. F. MCCULLOCH, J. L. BRAUN and H. C. ELLINGHAUSEN JR., *Nature*, Lond. 209, 939 (1966).
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